The azoreductase of yeast cells: a new feature of an old enzyme

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State of the art

Reduction of azo dyes removes colour

Yeasts are able to reduce azo dyes to amines

Reduction of polar azo dyes occurs extracellularly





From literature

In yeasts an externally directed redox enzyme system - located in the plasma membrane - is the ferric reductase

Ferric reductase activity is traditionally measured through the reduction of ferricyanide anion which is also impermeant to the plasma membrane

This same activity is also seen in the reduction of ferric citrate or Fe/EDTA complexes

Ferric reductase is involved in the iron uptake system of yeasts

Ferric reductase and iron uptake in yeasts

Yeasts have three systems to solubilize ferric iron from the environment:

- siderophores (Lesuisse et al., 2001);

- excretion of reducing agents or organic acids (like anthranilate and 3hydroxyanthranilate) (Bienfait *et al.*, 1987; Georgatsou *et al.*, 1994; Lesuisse *et al.* 1990);

- reduction to Fe²⁺ by a

membrane

<u>plasma</u>

reductase (Lesuisse et al., 1990; Dancis et al., 1990).

Iron Uptake in Yeast



ARN1-4: siderophore transporters FET4: low affinity ferrous iron transporter FET3+FTR1(permease): high affinity ferrous iron transporter ATX1: oxygen toxicity suppressor CCC2: iron transporter in the Golgi apparatus FIT: mannoproteins from the cell wall that retain iron

Methods

 Laboratorial Saccharomyces cerevisiae strain with the ability to reduce azo dyes

 Decolourization experiments along growth at 26°C and 120 rpm

 Activity assays of ferric and azo reductases with cells grown for 6h in normal decolourization medium

I. Measurement of both ferric and azo reductase activities along growth and decolourization:

Ferric and azo reductases have parallel activity curves with maxima in the late exponential growth phase



II. Addition of an known inhibitor of ferric reductase (IRON) to the decolourisation medium:



The addition of iron to the medium inhibits ferric both reductase at transcriptional postand transcriptional levels (Lesuisse et al. 1996) and delays decolorization

III. Deletion of the genes *FRE1* and *FRE2* in the strain: effect on azo and ferric reductase activities



The deletion of *FRE1* gene removes the decolorizing activity; *FRE2* removal does not affect the decolorizing ability, in our conditions

III. Deletion of the genes *FRE1* and *FRE2* in the strain: effect on decolourisation activity



Reduction of decolourising capacity by deletion of *FRE1* and *FRE2*

Fre1p is responsible for the major part of the azo reductase activity of intact yeast cells

IV. Use of inhibitors

- CCCP (carbonyl cyanide *m*-chlorophenylhydrazone)
 protonofore (Schröder *et al.*, 2003);
- DPI (diphenyleneiodonium chloride) flavoenzyme inhibitor (Wright and Kuhn 2002);
- Chloroquine antagonist of CoQ (Santos-Ocana et al., 1995);

• *p*-chloromercuribenzoate (pCMB) – thiol reagent susceptible of reacting with intra-membrane protein groups; inhibits NADH:ferricyanide reductase (Löw et al., 1978).

IV. Use of inhibitors



Conclusions

• The ferric reductase is the main activity involved in extracellular azo dye reduction by yeasts (fre1p mainly).

 However there is an alternative redox system to ferric reductase in the plasma membrane still unknown and that is able reduce some extracellular compounds (azo dyes).

Proposed model

Extracellular medium



• The major fraction of azo reductase activity depends on Fre1p [1; this work] and on a NADPH dehydrogenase [1] A

- Activity of Fre1p depends on a cytosolic NAD kinase [2] B
- Azo dye reduction (an presumably ferric iron reduction) must occur at an alternative site [this work] C

Proposed model

Extracellular medium



pCMB stimulates azo reductase activity at higher concentrations [this work]

• These observations are consistent with the existence of membrane transporters capable of switching electrons between two external reduction sites (electron or Q pool) D

[1] E Lesuisse, M Casteras-Simon and P Labbe 1996 JBC 271, 13578-13583 [2] S Kawai *et al.* 2001 FEMS Microbiol Lett 200, 181-184

Thanks for

your attention!