Synthesis Using Suzuki Cross Couplings of Sulfur Analogues of Dehydrotryptophan with a Definite Stereochemistry

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Keywords: Amino acids / Sulfur heterocycles / Suzuki coupling / Stereoisomerism

Sulfur analogues of dehydrotryptophan (**5–7**) were prepared in moderate to good yields (40–80%) by Suzuki cross coupling [Pd(PPh₃)₄, Na₂CO₃ or NaHCO₃, DME/H₂O, 90 °C] of several benzo[*b*]thiophene boronic acids with the methyl esters of *N-tert*-butyloxycarbonyl- β -bromodehydroalanine [Boc- Δ Ala(β -Br)-OMe] or *N-tert*-butyloxycarbonyl- β -bromodehydroaminobutyric acid [Boc- Δ Abu(β -Br)-OMe]. The β bromodehydroamino acid precursors **2** were, in turn, synthesized in high yields from the corresponding *N*,*N*-diacyldehydroamino acids **1** by treatment with trifluoroacetic acid (TFA)

Introduction

The insertion of non-proteinogenic conformationally constrained amino acids into bioactive peptides has proven to be a powerful tool in the elucidation of ligand-receptor binding interactions and in peptidomimetic design.^[1] Dehydroamino acids are among one of the most interesting classes of compounds that can be used for this purpose. These compounds are also found in several peptide antibiotics and are important substrates for the synthesis of novel amino acids through enantioselective catalytic hydrogenation.^[2] Although several methods have been described for their preparation, dehydroamino acids remain a challenging group of target molecules. Recently we have developed an efficient method for the stereoselective synthesis of the Zisomers of N,N-diacyldehydroamino acid derivatives from the corresponding β-hydroxyamino acids.^[3] These compounds have proved to be versatile substrates in Michael addition reactions, allowing the preparation of several new β-substituted amino acids and also β-substituted dehydroamino acids. By this method it was possible to obtain several tryptophan and dehydrotryptophan analogues using N-unprotected indoles as nucleophiles.^[4]

Here we report the use of our readily available N,N-diacyldehydroamino acid derivatives, after N-monodeprotection and β -bromination, in the Suzuki cross coupling reac-

 [a] Departamento de Química-Universidade do Minho 4700-320 Braga Portugal Fax: (internat.) +351-25/367-8983 E-mail: mjrpq@química.uminho.pt and N-bromosuccinimide (NBS) in two steps or in a one-pot procedure. Both procedures were stereoselective for the *E*isomer of Boc- Δ Ala(β -Br)-OMe. However, for Boc- Δ Abu(β -Br)-OMe, different ratios of *E*/*Z* isomers were obtained in each procedure. The stereoselectivity for the *Z*-isomer was increased greatly with the one-pot procedure. Pure isomers were used in the coupling reactions and the stereochemistry of the starting material was generally maintained. (© Wiley-VCH Verlag GmbH, 69451 Weinheim, Germany, 2002)

tion with benzo[b]thiophene derivatives to obtain new sulfur analogues of dehydrotryptophan. This coupling reaction was chosen because it is one of the most powerful and versatile methods for C-C bond formation as it: (i) tolerates a broad range of functionalities, (ii) can be performed under mild conditions, and (iii) usually maintains the stereochemistry of the starting material.^[5]

Results and Discussion

The Z-isomers of the methyl ester of N,N-diacyldehydroamino acids 1, obtained from the corresponding β -hydroxyamino acids,^[3] were N-monodeprotected and brominated in high yields to give β -bromo dehydroamino acids 2, using TFA and NBS either in two separate steps or in a one-pot procedure (Scheme 1). The latter gave slightly higher yields. In the case of Boc- Δ Ala(β -Br)-OMe, using both methods, only the E-isomer was obtained. Other authors have reported the formation of only the Z-isomer when the amine function is monoprotected either with acetyl^[6] or with methyloxycarbonyl^[7] groups. Our result could be due to the steric hindrance produced by the bulky Boc group. Mixtures of E/Z isomers were obtained in the synthesis of Boc- $\Delta Abu(\beta-Br)$ -OMe, with increased stereoselectivity for the Z-isomer, when the deprotection and bromination were carried out in a one-pot procedure (Scheme 1). This could result from the presence of TFA in the reaction mixture. The ratio of E/Z isomers obtained using the two step procedure (1:1) is in agreement with the bromination performed by

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Hoerrner et al.^[8] on the methyl ester of *N*-acetyldehydroaminobutyric acid. The isomers were separated by column chromatography and the stereochemistry was determined using differential NOE enhancements between the β (Δ Ala) or γ protons (Δ Abu) and the α NH.



Scheme 1

The pure β -bromodehydroamino acids (*E*)-**2a** and (*E*)- or (*Z*)-**2b** were prepared in order to be coupled under Suzuki coupling conditions with the commercially available benzo[*b*]thiophene 2- and 3-boronic acids and also with 7-boronic-2,3-dimethylbenzo[*b*]thiophene acid (**4**). The latter was prepared in order to study the reactivity of the benzene ring of the benzo[*b*]thiophene system, to which less attention has been paid. Compound **4** was obtained from 7-bromo-2,3-dimethylbenzo[*b*]thiophene (**3**) by halogen-lithium exchange, followed by boron transmetallation and treatment with acid (Scheme 2). Compound **3** was synthesized from 2-bromothiophenol and 3-bromobutan-2-one using 30% NaOH in the first step and polyphosphoric acid (PPA) in the cyclodehydration step (Scheme 2).^[9]



Scheme 2

The Suzuki coupling reactions were performed in DME/ water (4:1) using NaHCO₃ (2 equiv.), or in DME and aqueous 2 M Na₂CO₃ (2 equiv.).^[8] In both cases Pd(PPh₃)₄ (10 mol %) was used as the palladium catalyst at 90 °C for between 4 and 7.5 h (Scheme 3, Table 1).

The coupled products were obtained in moderate to good yields (Table 1); the corresponding deboronated benzo[*b*]-thiophenes were also isolated in small amounts (10-15% yield). The results show a similar reactivity of both rings of the benzo[*b*]thiophene system and of both β -bromo dehydroamino acid derivatives.

The stereochemistry of Boc- Δ Ala(β -Br)-OMe and Boc- Δ Abu(β -Br)-OMe was maintained, except in the case of the coupling of compound **4** with (*E*)-**2b**, in which (*E*)-**7b** and



Scheme 3. i) DME/H2O, Pd(PPh_3)_4 (10 mol %), Na_2CO_3 or NaHCO_3 (2 equiv.), 90 $^{\circ}\mathrm{C}$

Table 1. Yields of the coupled products

Benzo[b]thiophene	Δaa	Base, time	Product	Yield (%)
3-boronic acid	(<i>E</i>)- 2 a	Na ₂ CO ₃ , 4 h	(E)- 5 a	72
	(E)- 2b	Na_2CO_3 , 7 h	(E)- 5 b	52
	(Z)-2b	NaHCO ₃ , 5 h	(Z)-5b	62
2-boronic acid	(E)- 2 a	Na ₂ CO ₃ , 4 h	(E) -6a	81
	(<i>E</i>)-2b	NaHCO ₃ , 4 h	(<i>E</i>)-6b	61
	(Z)-2b	NaHCO ₃ , 6 h	(Z)-6b	47
4	(<i>E</i>)- 2 a	Na ₂ CO ₃ , 4 h	(<i>E</i>)-7a	40
	(E)-2b	NaHCO ₃ ,	(<i>E</i>)-7b	61
	< /	7 h 30 min		
			(Z)-7b	7
	(Z)- 2 b	NaHCO ₃ ,	(Z)-7b	70
	× /	5 h 30 min		

(Z)-7b were isolated in a ratio of 9:1. This could be due to base-catalysed isomerization^[10] to the thermodynamically more stable Z-isomer, since a longer reaction time was needed (7.5 h). With (Z)-2b as starting material, and under the same reaction conditions, no isomerization occurred and (Z)-7b was isolated in 70% yield after 5.5 h (Table 1). In both cases 2,3-dimethyl-7-hydroxybenzo[b]thiophene^[9] was also isolated in 11% yield.

As in the case of the starting materials, the stereochemistry of the coupled products was determined by NOE experiments. For dehydroaminobutyric acid derivatives the proton chemical shifts (in CDCl₃) of the methyl ester (OMe) and of the γ protons (Me) of the Z-isomers always appear at higher values than those of the E-isomers (Table 2).

Conclusion

The Suzuki coupling reaction was applied successfully to the synthesis of both E and Z isomers of several sulfur ana-

Table 2. 1H NMR chemical shifts (CDCl_3) of the OMe and Me groups of coupled ΔAbu derivatives

Bromination of the Methyl Esters of *N-tert*-butyloxycarbonyldehydroamino Acids

Compound	δ ΟΜε	δ Με
(E)-5b	3.34	2.19
(Z)-5b	3.90	2.30
(E)-6b	3.59	2.21
(Z)-6b	3.87	2.45
(E)-7b	3.37	2.21
(Z)-7b	3.89	2.21

logues of dehydrotryptophan. In this study the couplings were performed using benzo[*b*]thiophenes functionalized either on the thiophene or on the benzene ring. In general the stereochemistry of the β -bromo dehydroamino acid derivatives was maintained.

Both stereoisomers of the coupled products obtained could be useful in structure-activity relationship studies aimed at elucidating the biologically active conformation of peptides and also in the development of new molecules of pharmaceutical interest.

Experimental Section

General Remarks: Melting points were determined on a Gallenkamp apparatus and are uncorrected. The ¹H NMR spectra were measured on a Varian Unity Plus spectrometer at 300 MHz. Spinspin decoupling techniques were used to assign the signals. The ¹³C NMR spectra were measured on the same instrument at 75.4 MHz (using DEPT θ 45°). Elemental analyses were determined on a LECO CHNS 932 elemental analyser.

Column chromatography was performed on Macherey–Nagel silica gel 230–400 mesh. Petroleum ether refers to the boiling range 40-60 °C. When a solvent gradient was used the increase of polarity was done gradually from petroleum ether to mixtures of diethyl ether/petroleum ether increasing the amount of diethyl ether by 10% until the isolation of the product.

Preparation of the Methyl Esters of *N*,*N*-bis(*tert*-butyloxycarbonyl)-dehydroamino Acids

Synthesis of Boc₂-AAla-OMe (1a) and Boc₂-AAbu-OMe (1b): These compounds were prepared from Boc-Ser-OMe and Boc-Thr-OMe, respectively, using the *tert*-butylpyrocarbonate/dimethylaminopyridine procedure described elsewhere.^[3]

Monodeprotection of the Methyl Esters of *N*,*N*-bis(*tert*-butyloxycarbonyl)dehydroamino Acids

Synthesis of Boc-AAla-OMe and Boc-AAbu-OMe: Compound 1a (1.51 g, 5 mmol) or 1b (1.58 g, 5 mmol) was dissolved in dichloromethane (0.1 M) and 2% of TFA was added slowly with vigorous stirring. The reaction was monitored by TLC and when no starting material was detected (ca. 1 hour) an additional 50 mL of dichloromethane was added. The organic phase was then washed with NaHCO₃ (1 M) and brine (3 × 30 mL each). After drying over MgSO₄ the extract was taken to dryness at reduced pressure to afford Boc- Δ Ala-OMe (0.98 g, 98%), as an oil,^[3] or Boc- Δ Abu-OMe (0.96 g, 89%). Crystallization of the latter from diethyl ether/*n*-hexane gave the pure *Z*-isomer as a white solid; m.p. 69.5–71.0 °C. Synthesis of Boc-(*E*)-ΔAla(β-Br)-OMe [(*E*)-2a]: Boc-ΔAla-OMe (1.01 g, 5 mmol) was dissolved in dichloromethane (0.1 M) and 1.2 equiv. of *N*-bromosuccinimide were added with vigorous stirring. After reacting for 16 hours, triethylamine (1.5 equiv.) was added and stirring continued for an additional hour. The solvent was then evaporated at reduced pressure and the residue partitioned between 100 mL of dichloromethane and 50 mL of KHSO₄ (1 M). The organic phase was washed with KHSO₄ (1 M), NaHCO₃ (1 M) and brine (3 × 30 mL each). After drying over MgSO₄ the extract was taken to dryness at reduced pressure to afford (*E*)-2a (1.15 g, 82%) as a colourless oil. ¹H NMR (CDCl₃): $\delta = 1.47$ (s, 9 H, CH₃ Boc), 3.82 (s, 3 H, OCH₃), 6.19 (broad s, 1 H, αNH), 6.87 (s, 1 H, βCH) ppm. ¹³C NMR (CDCl₃): $\delta = 27.78$ [C(CH₃)₃], 52.78 (OCH₃), 81.61 [OC(CH₃)₃], 109.17 (CH), 132.50 (C), 151.83 (C=O), 162.70 (C=O) ppm.

Synthesis of Boc- Δ Abu(β -Br)-OMe [(*E*)-2b and (*Z*)-2b]: The same procedure as described above was followed, substituting Boc- Δ Ala-OMe for Boc- Δ Abu-OMe (1.08 g, 5 mmol) to give a 1:1 mixture of (*E*)-2b and (*Z*)-2b (1.35 g, 92%). The diastereomers were separated by column chromatography using 30% diethyl ether/petroleum ether as eluent.

(*E*)-2b: M.p. 108.0–109.0 °C (from diethyl ether/*n*-hexane). ¹H NMR (CDCl₃): $\delta = 1.46$ (s, 9 H, CH₃ Boc), 2.41 (s, 3 H, γ CH₃), 3.82 (s, 3 H, OCH₃), 6.00 (broad s, 1 H, α NH) ppm. ¹³C NMR (CDCl₃): $\delta = 25.54$ (C=CCH₃), 28.02 [C(CH₃)₃], 52.19 (OCH₃), 81.21 [OC(CH₃)₃], 121.95 (C), 126.23 (C), 152.63 (C=O), 164.45 (C=O) ppm. C₁₀H₁₆BrNO₄ (294.1): calcd. C 40.83, H 5.44, N 4.76; found C 40.92, H 5.43, N 4.88.

(*Z*)-2b: M.p. 75.5–77.0 °C (from *n*-hexane). ¹H NMR (CDCl₃): $\delta = 1.46$ (s, 9 H, CH₃ Boc), 2.52 (s, 3 H, γ CH₃), 3.83 (s, 3 H, OCH₃), 6.23 (broad s, 1 H, α NH) ppm. ¹³C NMR (CDCl₃): $\delta = 24.43$ (C=CCH₃), 28.00 [C(CH₃)₃], 52.42 (OCH₃), 81.46 [OC(CH₃)₃], 120.40 (C), 127.33 (C), 152.31 (C=O), 163.17 (C=O) ppm. C₁₀H₁₆BrNO₄ (294.1): calcd. C 40.83, H 5.44, N 4.76; found C 41.19, H 5.52, N 4.81.

One-Pot Procedure for Synthesis of the β-Bromo *N-tert*-butyloxycarbonyldehydroamino Acid Methyl Esters

Synthesis of Boc-(*E*)- Δ Ala(β -Br)-OMe [(*E*)-2a]: Boc₂- Δ Ala-OMe (1.51 g, 5 mmol) was dissolved in dichloromethane (0.1 M) and 2% of TFA was added slowly with vigorous stirring. The reaction was monitored by TLC and when no starting material was detected (ca. 1 hour) 1.2 equiv. of *N*-bromosuccinimide was added. After reacting for 16 hours triethylamine (3 equiv.) was added and stirring continued for an additional hour. The solvent was then evaporated at reduced pressure and the product (*E*)-2a (1.25 g, 89%) was isolated as an oil, according to the procedure given above.

Synthesis of Boc- Δ Abu(β -Br)-OMe [(*E*)-2b and (*Z*)-2b]: The same procedure as described above was followed, substituting Boc₂- Δ Ala-OMe for Boc₂- Δ Abu-OMe (1.58 g, 5 mmol) to give (*E*)-2b and (*Z*)-2b (1.31 g, 89%) as a 1:9 mixture. Separation of the diastereomers was carried out by column chromatography using 30% diethyl ether/petroleum ether as eluent. Alternatively, (*Z*)-2b could be obtained by crystallization from diethyl ether/*n*-hexane.

Synthesis of 7-Boronic-2,3-dimethylbenzo[b]thiophene Acid (4)

7-Bromo-2,3-dimethylbenzo[*b***]thiophene (3):** A 30% aqueous NaOH solution (7 mL, 53 mmol) was added with stirring to 2-bromothi-ophenol (4.90 g, 26 mmol) at 0 °C. After 30 min 3-bromobutan-2-

one (3.70 g, 26 mmol) was added dropwise from an addition funnel and the mixture was left stirring for 30 min at room temp. Chloroform (30 mL) and water (20 mL) were then added and the phases separated. The aqueous phase was extracted with chloroform (2 imes30 mL) and the combined organic phases were dried (MgSO₄), filtered and the solvent removed to give the corresponding ketoarylsulfide as a light yellow oil (5.04 g, 75%). ¹H NMR (CDCl₃): $\delta =$ 1.45 (d, J = 7.0 Hz, 3 H, SCHCH₃), 2.20 (s, 3 H, COCH₃), 3.85 $(q, J = 7.0 \text{ Hz}, 1 \text{ H}, \text{SCHCH}_3), 7.04 \text{ (broad t}, J = 7.5 \text{ Hz}, 1 \text{ H}, 4-$ H), 7.20 (broad t, J = 7.5 Hz, 1 H, 5-H), 7.32 (broad d, J = 7.5 Hz, 1 H, 6-H), 7.22 (broad d, J = 7.5 H, 1 Hz, 3-H). This compound (5.00 g, 20 mmol) was added dropwise from an addition funnel to PPA (10 g) which had previously been heated at 150 °C for 15 min. The mixture was then heated for 1 h at 150 °C. After cooling, crushed ice was added (30 g) and it was left stirring until the ice had melted. Chloroform was added (30 mL) and the phases were separated. The aqueous phase was extracted with chloroform (2 \times 50 mL) and the organic phase was dried (MgSO₄), filtered and the solvent removed to give compound 3 (4.80 g, quantitative yield), m.p. 67.0–68.5 °C (from *n*-hexane). ¹H NMR (CDCl₃): $\delta = 2.29$ (s, 3 H, Ar-CH₃), 2.51 (s, 3 H, Ar-CH₃), 7.23 (t, J = 8.0 Hz, 1 H, 5-H), 7.42 (dd, *J* = 8.0, 1.0 Hz, 1 H, 4-H), 7.54 (dd, *J* = 8.0, 1.0 Hz, 1 H, 6-H) ppm. ¹³C NMR (CDCl₃): $\delta = 11.78$ (CH₃), 13.77 (CH₃), 115.46 (C), 120.00 (CH), 125.16 (CH), 126.23 (CH), 132.88 (C), 134.99 (C), 139.61 (C), 142.02 (C) ppm. C₁₀H₉BrS (241.1): calcd. C 49.81, H 3.76, S 13.29; found C 49.40, H 3.86, S 13.30.

7-Boronic-2,3-dimethylbenzo[b]thiophene acid (4): A solution 1.6 M of *n*BuLi in hexane (5.1 mL, 8.1 mmol) was added dropwise to a solution of bromo compound 3 (1.50 g, 6.2 mmol) in dry ether (35 mL) at 0 °C, under argon. A white precipitate was formed and the mixture was left stirring for 20 min at 0 °C. The temperature was lowered to -20 °C, tributyl borate was added (2.2 mL, 8.1 mmol), and the mixture was stirred at this temperature for 10 min, becoming a limpid solution that was left stirring for 2 h at room temp. The mixture was then cooled to 0 °C and acidified to pH 1-2 with 1 N HCl. More ether was added and the mixture stirred for 15 min. The phases were separated, and a 20% aqueous solution of NaOH (50 mL) was added to the organic phase to form a white precipitate. The two phases were filtered under vacuum and a white solid was separated. This solid was completely dissolved with 1 N HCl (20 mL) and the mixture was extracted with ether (3 \times 20 mL). The combined organic phases were dried (MgSO₄), filtered and solvent removal gave compound 4 as a white solid (0.60 g, 47%), m.p. 288.0–290.0 °C. ¹H NMR ([D₆]DMSO): $\delta = 2.24$ (s, 3 H, Ar-CH₃), 2.42 (s, 3 H, Ar-CH₃), 7.31 (apparent t, J = 7.0 Hz, 1 H, 5-H), 7.65 (broad d, J = 7.0 Hz, 1 H, 4 or 6-H), 7.65 (broad d, J = 7.0 Hz, 1 H, 4 or 6-H), 8.30 (broad s, 2 H, 2 × OH) ppm. ¹³C NMR ([D₆]DMSO): $\delta = 11.07$ (CH₃), 13.33 (CH₃), 109.30 (C), 123.04 (CH), 123.14 (CH), 125.61 (C), 130.15 (CH), 134.33 (C), 140.23 (C), 143.16 (C) ppm.

General Procedure for Palladium-Catalysed Cross Couplings of Benzo[*b*]thiophene Boronic Acids with β -Bromo Dehydroamino Acid Derivatives: Compounds (*E*)-2a, (*E*)-2b or (*Z*)-2b (1 mmol) were coupled with benzo[*b*]thiophene boronic acids (1.1 equiv.) in a mixture of either DME/water (4 mL/1 mL) and NaHCO₃ (2 equiv.) or DME (5 mL) and aqueous 2 M Na₂CO₃ (2 equiv.), and Pd(PPh₃)₄ (10 mol %). The reactions were followed by TLC which determined the different reaction times (Table 1). After cooling, diethyl ether and water were added and the phases were separated. The aqueous phase was washed with diethyl ether (3 × 50 mL). The combined organic phases were dried (MgSO₄), filtered and solvent removal gave a brown residue which was submitted to column chromatography affording the coupled products. The corresponding deboronated benzo[b]thiophenes were isolated as the less polar products in 10-15% yield.

Boc-(*E*)-ΔAla[β-(benzo[*b*]thiophen-3-yl)]-OMe [(*E*)-5a)]: Column chromatography using a solvent gradient from neat petroleum ether to 40% diethyl ether/petroleum ether gave compound (*E*)-5a as a white solid (0.24 g, 72%), m.p. 158.0–159.0 °C. ¹H NMR (CDCl₃): $\delta = 1.41$ (s, 9 H, CH₃ Boc), 3.91 (s, 3 H, OCH₃), 6.17 (s, 1 H, α-NH), 7.40–7.47 (m, 2 H, 2 × Ar-H), 7.58 (s, 1 H, 2-H), 7.86–7.92 (m, 3 H, 2 × Ar-H and β-CH) ppm. ¹³C NMR (CDCl₃): $\delta = 28.07$ [C(CH₃)₃], 52.65 (OCH₃), 81.14 [OC(CH₃)₃], 121.61 (CH), 121.74 (CH), 122.72 (CH), 124.65 (CH), 124.92 (CH), 125.31(C), 128.28 (CH), 129.19 (C), 138.24 (C), 139.20 (C), 153.21 (C=O), 165.91 (C=O) ppm. C₁₇H₁₉NO₄S (333.4): calcd. C 61.24, H 5.74, N 4.20, S 9.62; found C 61.12, H 5.77, N 4.23, S 9.52.

Boc-(*E*)-**ΔAla**[**β**-(benzo]*b*]thiophen-2-yl)]-OMe [(*E*)-6a]: Column chromatography using a solvent gradient from neat petroleum ether to 40% diethyl ether/petroleum ether gave compound (*E*)-6a as a white solid (0.27 g, 81%), m.p. 150.0–151.0 °C. ¹H NMR (CDCl₃): $\delta = 1.50$ (s, 9 H, CH₃ Boc), 3.87 (s, 3 H, OCH₃), 5.98 (s, 1 H, α-NH), 7.25–7.44 (m, 2 H, 2 × Ar-H), 7.55 (s, 1 H, 3-H), 7.74 (broad s, 1 H, β-CH), 7.77–7.86 (m, 2 H, 2 × Ar-H) ppm. ¹³C NMR (CDCl₃): $\delta = 28.20$ [C(CH₃)₃], 52.53 (OCH₃), 81.25 [OC(CH₃)₃], 122.21 (CH), 124.28 (CH), 124.69 (CH), 125.91 (CH), 128.52 (CH), 129.73 (CH), 134.99 (C), 136.67 (C), 138.29 (C), 141.83 (C), 153.60 (C=O), 165.51 (C=O) ppm. C₁₇H₁₉NO₄S (333.4): calcd. C 61.24, H 5.74, N 4.20, S 9.62; found C 61.00, H 5.74, N 4.07, S 9.31.

Boc-(*E***)-ΔAla[β-(2,3-dimethylbenzo[***b***]thiophen-7-yl)]-OMe [(***E***)-7a]: Column chromatography using a solvent gradient from neat petroleum ether to 40% diethyl ether/petroleum ether gave the product as a white solid (0.14 g, 40%), m.p. 125.0–126.0 °C. ¹H NMR (CDCl₃): \delta = 1.34 (s, 9 H, CH₃ Boc), 2.31 (s, 3 H, Ar-CH₃), 2.50 (s, 3 H, Ar-CH₃), 3.90 (s, 3 H, OCH₃), 6.25 (s, 1 H, α-NH), 7.33–7.39 (m, 2 H, β-CH and 5-H), 7.50–7.60 (m, 2 H, 4- and 6-H) ppm. ¹³C NMR (CDCl₃): \delta = 11.45 (CH₃), 13.74 (CH₃), 27.99 [C(CH₃)₃], 52.67 (OCH₃), 80.99 [OC(CH₃)₃], 121.71 (CH), 123.47 (CH), 124.00 (CH), 125.28 (CH), 126.23 (C), 127.43 (C), 128.19 (C), 133.79 (C), 138.47 (C), 141.49 (C), 152.45 (C=O), 165.77 (C= O) ppm. C₁₉H₂₃NO₄S (361.5): calcd. C 63.14, H 6.41, N 3.88, S 8.87; found C 63.39, H 6.57, N 3.88, S 8.68.**

Boc-(*Z***)-**Δ**Abu**[β-(benzo[*b*]thiophen-2-yl)]-OMe [(*E*)-6b]: Column chromatography using a solvent gradient from neat petroleum ether to 30% diethyl ether/petroleum ether gave product (*E*)-6b (0.21 g, 61%). Crystallization from diethyl ether/*n*-hexane gave white crystals, m.p. 116.0–118.0 °C. ¹H NMR (CDCl₃): δ = 1.49 (s, 9 H, CH₃ Boc), 2.21 (s, 3 H, γ-CH₃), 3.59 (s, 3 H, OCH₃), 6.14 (s, 1 H, α-NH), 7.14 (s, 1 H, 3-H), 7.30–7.41 (m, 2 H, 2 × Ar-H), 7.68–7.82 (m, 2 H, 2 × Ar-H) ppm. ¹³C NMR (CDCl₃): δ = 21.00 (C=CCH₃), 28.14 [C(CH₃)₃], 52.26 (OCH₃), 81.38 [OC(CH₃)₃], 122.07 (CH), 122.53 (CH), 123.54 (CH), 124.35 (CH), 124.43 (CH), 125.50 (C), 126.55 (C), 139.46 (C), 139.96 (C), 142.62 (C), 152.67 (C=O), 165.95 (C=O) ppm. C₁₈H₂₁NO₄S (347.4): calcd. C 62.23, H 6.09, N 4.03, S 9.23; found C 62.29, H 6.10, N 4.00, S 8.99.

Boc-(*Z*)-**ΔAbu-[β-(benzo[***b***]thiophen-2-yl)]-OMe [(***Z***)-6b]: Column chromatography using a solvent gradient from neat petroleum ether to 40% diethyl ether/petroleum ether gave (***Z***)-6b as a white solid (0.16 g, 47%). Crystallization from diethyl ether/***n***-hexane gave white crystals m.p.108.0–110.0 °C. ¹H NMR (CDCl₃): \delta = 1.48 (s, 9 H, CH₃ Boc), 2.45 (s, 3 H, γ-CH₃), 3.87 (s, 3 H, OCH₃), 6.20 (s, 1 H, α-NH), 7.32–7.42 (m, 2 H, 2 × Ar-H), 7.49 (s, 1 H, 3-H), 7.76–7.86 (m, 2 H, 2 × Ar-H) ppm. ¹³C NMR (CDCl₃): \delta = 19.37**

 $\begin{array}{l} (C=CCH_3), \ 28.19 \ [C(CH_3)_3], \ 52.21 \ (OCH_3), \ 81.21 \ [OC(CH_3)_3], \\ 121.97 \ (CH), \ 123.94 \ (CH), \ 124.57 \ (CH), \ 125.28 \ (CH), \ 129.06 \ (C), \\ 138.79 \ (C), \ 140.53 \ (C), \ 140.98 \ (C), \ 153.38 \ (C=O), \ 165.90 \ (C=O) \\ ppm. \ C_{18}H_{21}NO_4S \ (347.4): \ calcd. \ C \ 62.23, \ H \ 6.09, \ N \ 4.03, \ S \ 9.23; \\ found \ C \ 62.29, \ H \ 6.10, \ N \ 4.00, \ S \ 8.99. \end{array}$

Boc-(*E***)-ΔAbu[β-(benzo[***b***]thiophen-3-yl)]-OMe [(***E***)-5b]: Column chromatography using a solvent gradient from neat petroleum ether to 20% diethyl ether/petroleum ether gave the product as a white solid (0.18 g, 52%), m.p. 120.0–122.0 °C. ¹H NMR (CDCl₃): \delta = 1.51 (s, 9 H, CH₃ Boc), 2.19 (s, 3 H, γ-CH₃), 3.34 (s, 3 H, OCH₃), 6.17 (s, 1 H, α-NH), 7.19 (s, 1 H, 2-H), 7.30–7.40 (m, 2 H, 2 × Ar-H), 7.62–7.70 (m, 1 H, Ar-H) 7.82–7.88 (m, 1 H, Ar-H) ppm. ¹³C NMR (CDCl₃): \delta = 20.93 (C=CCH₃), 28.19 [C(CH₃)₃], 51.84 (OCH₃), 80.95 [OC(CH₃)₃], 122.45 (CH), 122.64 (CH), 124.19 (CH), 124.31 (CH), 126.23 (C), 132.40 (C), 136.50 (C), 137.81 (C), 139.67 (C), 153.06 (C=O), 165.44 (C=O) ppm. C₁₈H₂₁NO₄S (347.4): calcd. C 62.23, H 6.09, N 4.03, S 9.23; found C 62.34, H 6.19, N 4.07, S 9.17.**

Boc-(*Z***)-**Δ**Abu**[β-(benzo]*b*]thiophen-3-yl)]-OMe [(*Z*)-5b]: Column chromatography using a solvent gradient from neat petroleum ether to 30% diethyl ether/petroleum ether gave the product as a white solid (0.22 g, 62%) m.p. 122.0–124.0 °C. ¹H NMR (CDCl₃): $\delta = 1.39$ (s, 9 H, CH₃ Boc), 2.30 (s, 3 H, γ-CH₃), 3.90 (s, 3 H, OCH₃), 5.65 (s, 1 H, α-NH), 7.32–7.43 (m, 3 H, 3 × Ar-H), 7.58–7.65 (m, 1 H, Ar-H) 7.86–7.93 (m, 1 H, Ar-H) ppm. ¹³C NMR (CDCl₃): $\delta = 20.74$ (C=CCH₃), 27.99 [C(CH₃)₃], 52.03 (OCH₃), 80.67 [OC(CH₃)₃], 122.74 (CH), 124.50 (CH), 124.72 (CH), 125.85 (C), 127.87 (C), 135.37 (C), 136.17 (C), 140.03 (C), 153.02 (C=O), 165.45 (C=O) ppm. C₁₈H₂₁NO₄S (347.4): calcd. C 62.23, H 6.09, N 4.03, S 9.23; found C 62.05, H 6.21, N 4.00, S 9.08.

Boc-(*Z***)-**Δ**Abu**[β-(2,3-dimethylbenzo[*b*]thiophen-7-yI)]-OMe [(*Z*)-7b]: Column chromatography using a solvent gradient from neat petroleum ether to 30% diethyl ether/petroleum ether gave the product as a white solid (0.26 g, 70%), m.p. 116.0–118.0 °C. ¹H NMR (CDCl₃): $\delta = 1.35$ (s, 9 H, CH₃ Boc), 2.27 (s, 3 H, γ-CH₃), 2.32 (s, 3 H, Ar-CH₃) 2.47 (s, 3 H, Ar-CH₃), 3.89 (s, 3 H, OCH₃), 5.70 (s, 1 H, *α*-NH), 7.10 (broad signal, 1 H, 6-H), 7.39 (apparent t, *J* = 7.6 Hz, 1 H, 5-H) 7.56 (broad d, *J* = 7.9 Hz, 1 H, 4-H) ppm. ¹³C NMR (CDCl₃): $\delta = 11.49$ (CH₃), 13.71 (CH₃), 19.63 (C= CCH₃), 28.04 [C(CH₃)₃], 52.08 (OCH₃), 80.60 [OC(CH₃)₃], 120.74 (CH), 122.53 (CH), 124.52 (CH), 125.24 (C), 127.49 (C), 131.41 (C), 133.62 (C), 134.35 (C), 135.82 (C), 141.57 (C), 153.03 (C=O), 165.56 (C=O) ppm. C₂₀H₂₅NO₄S (375.5): calcd. C 63.98, H 6.71, N 3.73, S 8.54; found C 63.65, H 6.78, N 3.84, S 8.44.

7-Hydroxy-2,3-dimethylbenzo[b]thiophene^[9] (0.02 g, 11%) was also isolated, eluting from the column after the deboronated compound and before the coupled product.

Boc-(*E*)- Δ Abu-[β -(2,3-dimethylbenzo[*b*]thiophen-7-yl)]-OMe I(E)-7b]: Column chromatography using a solvent gradient from neat petroleum ether to 30% diethyl ether/petroleum ether gave (Z)-7b as a white solid (0.03 g, 7%), followed by (E)-7b, also as a white solid (0.23 g, 61%) m.p. 116.0–118.0 °C. ¹H NMR (CDCl₃): δ = 1.51 (s, 9 H, CH₃ Boc), 2.21 (s, 3 H, γ-CH₃), 2.29 (s, 3 H, Ar-CH₃), 2.46 (s, 3 H, Ar-CH₃), 3.37 (s, 3 H, OCH₃), 6.12 (s, 1 H, α-NH), 7.03 (broad d, J = 7.3 Hz, 1 H, 6-H), 7.33 (apparent t, J = 7.9, 7.3 Hz, 1 H, 5-H), 7.50 (dd, J = 7.9, 0.9 Hz, 1 H, 4-H) ppm. ¹³C NMR (CDCl₃): $\delta = 11.48$ (CH₃), 13.71 (CH₃), 20.50 (C=CCH₃), 28.21 [C(CH₃)₃], 51.77 (OCH₃), 80.79 [OC(CH₃)₃], 120.10 (CH), 122.01 (CH), 123.93 (CH), 125.12 (C), 127.22 (C), 133.66 (C), 135.51 (C), 136.15 (C), 139.01 (C), 141.12 (C), 153.09 (C=O), 164.95 (C=O) ppm. C₂₀H₂₅NO₄S (375.5): calcd. C 63.98, H 6.71, N 3.73, S 8.54; found C 63.87, H 6.79, N 3.76, S 8.47.

7-Hydroxy-2,3-dimethylbenzo[b]thiophene^[9] (0.02 g, 11%) was also isolated, eluting from the column after the deboronated compound and before the coupled products.

Acknowledgments

The Foundation for Science and Technology (Portugal) is thanked for financial support through IBQF-Univ. Minho, POCTI/99/QUI/ 32689 project (also financial support of N.O. Silva) and through SFRH/BD/4709/2001 (PhD financial support of A.S. Abreu).

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Received March 6, 2002 [O02119]