# Chemistry SELECT 

## Supporting Information

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## Synthesis and Bioactivity of New Analogue of Bicyclic 1Azafagomine

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## Author Contributions

M.K. Supervision:Lead
A.H. Investigation:Equal

## SUPPORTING INFORMATION

## EXPERIMENTAL

## General chemistry

Solvents were distilled under anhydrous conditions. All reagents were purchased and used without further purification. Glassware was dried prior to use. Compounds were purified by dry flash chromatography using silica $60,<0063 \mathrm{~mm}$ and water pump vacuum or by flashchromatography using silica $60 \AA$ 230-400 mesh as stationary phases. TLC plates (silica gel 60 $\mathrm{F}_{254}$ ) were visualized either at a UV lamp or in iodine. Melting points are uncorrected. The specific rotation was determined by AUTOPOL III polarimeter.

Elemental analysis was performed on the analyzer Carlo Erba 1108.
NMR experiments: The NMR experiments have been performed on a BRUKER FT NMR spectrometer AVANCE 400 (Bruker, Karlsruhe, Germany) ( 400 MHz for ${ }^{1} \mathrm{H}$ and 100.6 MHz for ${ }^{13} \mathrm{C}$ ) with a BVT 3200 variable temperature unit in 5 mm sample tubes using Bruker Standard software (TopSpin 3.1). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical shifts were referenced to internal tetramethylsilane (TMS); the experimental parameters for ${ }^{1} \mathrm{H}$ are as follows: digital resolution $=$ $0.23 \mathrm{~Hz}, \mathrm{SWH}$ (spectral width in Hz ) $=8224 \mathrm{~Hz}$, TD (time domain) $=65 \mathrm{~K}$, SI (Fourier transform size) $=32 \mathrm{~K}, 90$ pulse-length $=10 \mu \mathrm{~s}$, PL1 (power level for F1 channel) $=3 \mathrm{~dB}$, ns (number of scans) $=1$, ds (number of dummy scans) $=0, \mathrm{~d} 1$ (relaxation delay) $=1 \mathrm{~s}$ and for ${ }^{13} \mathrm{C}$ as follows: digital resolution $=0.27 \mathrm{~Hz}, \mathrm{SWH}=25253 \mathrm{~Hz}, \mathrm{TD}=65 \mathrm{~K}, \mathrm{SI}=32 \mathrm{~K}, 90$ pulse-length $=9 \mu \mathrm{~s}, \mathrm{PL} 1=2$ $d B, n s=100, d s=2, d 1=3 \mathrm{~s}$.

HSQC: pulse program $=$ hsqcetgp, digital resolution $=1.97 \mathrm{~Hz}, \mathrm{SWH}=5342, \mathrm{TD}=4096, \mathrm{SI}$ $=1024,90$ pulse-length $=20 \mu \mathrm{~s}, \mathrm{PL1}=3 \mathrm{~dB}, \mathrm{~ns}=4, \mathrm{ds}=16, \mathrm{~d} 1=1.5 \mathrm{~s}$.

HMBC: pulse program $=$ hmbcgplpndgf, digital resolution $=1.97 \mathrm{~Hz}, \mathrm{SWH}=6810, \mathrm{TD}=$ $4096, \mathrm{SI}=2048,90$ pulse-length $=20 \mu \mathrm{~s}, \mathrm{PL} 1=3 \mathrm{~dB}, \mathrm{~ns}=24, \mathrm{ds}=16, \mathrm{~d} 1=1.5 \mathrm{~s}$.

The NMR-grades DMSO- $\mathrm{d}_{6}, \mathrm{CDCl}_{3}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{D}_{2} \mathrm{O}$ was used for the solutions of 1-7.
IR experiments: Infrared spectra were recorded on a Bomem MB 104. Samples were run as oils as thin films. The spectrum was taken in the range of $4000-400 \mathrm{~cm}^{-1}$ at room temperature.

MS experiments: MS spectra were recorded on a Varian 500-MS LC Ion Trap Mass and VG Autopsic M spectrometer.

## Synthesis of new analogue of bicyclic 1-azafagomine (1-7)

Synthesis of (E)-2,4-pentadienol (1)

2,4-Pentadienol 1 was obtained by repeating the literature procedure. In details, 250 ml of a necked bottom flask with a magnetic stir bar was placed in an ice bath, where 5.09 g of 2,4pentadienic acid, which was dissolved in dried THF ( 77 ml ) and $8.9 \mathrm{ml}^{\text {of }} \mathrm{NEt}_{3}$ were added. Then 5 ml of ethylchloroformate was dissolved in 13 ml of THF, placed in dropping funnel, connected with 250 ml three-bottom flask and left it to pour drop by drop for 30 min at $-5^{\circ}-0^{\circ} \mathrm{C}$. After 30 min reaction was stopped, precipitate filtrated under vacuum and washed thoroughly with THF ( $3 \times 50 \mathrm{ml}$ ) (Solution A).

A 1-I three-necked round bottom flask was placed in an ice bath $\left(7^{\circ} \mathrm{C}\right)$ where 515 ml of $\mathrm{H}_{2} \mathrm{O}$ and 4.804 g of $\mathrm{NaBH}_{4}$ were added and stirred. Solution A was then added dropwise over 30 min after which the solution was removed from the ice bath and allowed to stir at ambient temperature for 3.5 h . Thereafter, the solution was reinserted into an ice-water bath and 20 ml of $37 \% \mathrm{HCl}$ was added dropwise using a dropping funnel under an $\mathrm{N}_{2}$ atmosphere. The mixture obtained was extracted with diethyl ether ( $2 \times 100 \mathrm{ml}$ ), and the organic layers were combined and extracted with $10 \% \mathrm{NaOH}$ solution ( 50 ml ), followed by extraction with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{ml})$ and brine ( 50 ml ), then dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated to dryness under reduced pressure to give a colorless liquid in $31.4 \%$ yield, b.p. $58-62^{\circ} \mathrm{C} / 11 \mathrm{mmHg}(15 \mathrm{~Pa})$. Yield $31.4 \%$. The spectroscopic data corresponded to literature values [46], $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3616,3444$, 2873, 2254, 1824, 1731, 1606, 1455, 1377, 1083, 994, 902; $\delta_{H}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 6.25(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2}=\mathrm{CH}$ ), $5.74\left(1 \mathrm{H}, \mathrm{dt}, \mathrm{J}=14.7,5.7, \mathrm{CH}=\mathrm{CH}_{2} \mathrm{OH}\right), 5.13(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=16.4, \mathrm{CH}), 5.01(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=9.6, \mathrm{CH})$, 4.07 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}$ ), 3.12 ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}$ ); $\delta \mathrm{C}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 136.2$ (CH), 132.5 (CH), 131.2 $(\mathrm{CH}), 116.9\left(\mathrm{CH}_{2}\right), 62.3\left(\mathrm{CH}_{2}\right)$. Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 71.42 ; \mathrm{H}, 9.52$. Found: $\mathrm{C}, 71.47 ; \mathrm{H}, 9.48$.

## Synthesis of (S)-Diethyl-3-(hydroxymethyl)pyridazine-1,2(3H,6H)-dicarboxylate (2)

Compounds 2 was obtained according to the procedure. In general, in order to obtain compound 2, two different solutions (A and B) were prepared.

Solution A: a solution of $\mathrm{Me}_{2} \mathrm{Zn} 1.2 \mathrm{M}$ in toluene ( $991 \mu \mathrm{l} ; 1.19 \mathrm{mmol}$ ) was added to a solution of penta-2,4-dien-1-ol $1(0.100 \mathrm{~g} ; 1.19 \mathrm{mmol})$ in dry toluene ( 6 ml ) at $0^{\circ} \mathrm{C}$, and stirred for 5 min .

Solution B: a solution of $\operatorname{MeMgBr} 1.4 \mathrm{M}$ in toluene/THF ( $849 \mu \mathrm{~L} ; 1.19 \mathrm{mmol}$ ) was added to a solution of $(S)$-BINOL ( $0.340 \mathrm{~g} ; 1.19 \mathrm{mmol}$ ) in dry toluene ( 6 ml ) at $0^{\circ} \mathrm{C}$, and stirred for 5 min .

Thereafter, solution A was diluted in dry toluene ( 10 ml ), added to solution B, stirred for 5 min and then refrigerated at $-78 \circ \mathrm{C}$. To this mixture, a solution of diethyl azodicarboxylate ( $543 \mu \mathrm{~L} ; 1.19 \mathrm{mmol}$ ) in dry toluene ( 10 ml ) was added. The temperature was allowed to rise
gradually to room temperature and the reaction mixture was stirred for 18 h . The reaction was quenched with a saturated solution of $\mathrm{NaHCO}_{3}(1 \mathrm{ml})$, filtered through a pad of Celite ${ }^{\oplus}$, which was washed with EtOAc ( $3 \times 20 \mathrm{ml}$ ). The filtrates were combined and concentrated under reduced pressure giving a yellow oil. The crude oil was purified by "dry-flash" chromatography (silica, petroleum ether / diethyl ether). (S)-BINOL was recovered with petroleum ether (1): diethyl ether (1) ( $0.200 \mathrm{~g} ; 69 \%$ ) and product ( - )-2 with diethyl ether ( $0.225 \mathrm{~g} ; 73 \%$ ) as a yellow oil. $[\alpha]_{D}{ }^{20}-23.40$ (conc. $1.25 \%$ in $\mathrm{CHCl}_{3}$ ). $v_{\max }$ (neat) $3483,1707 \mathrm{~cm}^{-1}$. $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.23-$ $1.30\left(12 \mathrm{H}, \mathrm{m}, 4 \mathrm{CH}_{3}, \mathrm{~A}+\mathrm{B}\right), 2.58(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.35\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=12.3,9.5 \mathrm{~Hz}, \mathrm{H}-3^{\prime}, \mathrm{A}\right), 3.45(1 \mathrm{H}, \mathrm{dd}$, $\left.J=12.0,9.8 \mathrm{~Hz}, \mathrm{H}-3^{\prime}, B\right), 3.56-3.69\left(2 \mathrm{H}, \mathrm{m}, 2 \mathrm{H}-3^{\prime}, \mathrm{A}+\mathrm{B}\right), 3.77(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=13.5,4.3 \mathrm{~Hz}, \mathrm{H}-6, \mathrm{~A}), 3.91$ (1H, br s, H-6, B), 4.11-4.26 ( $8 \mathrm{H}, \mathrm{m}, 4 \mathrm{CH}_{2}, \mathrm{~A}+\mathrm{B}$ ), $4.30(1 \mathrm{H}, \mathrm{tdd}, \mathrm{J}=6.0,3.9,2.2 \mathrm{~Hz}, \mathrm{H}-6, B), 4.34-$ $4.44(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6, \mathrm{~A}), 4.72(2 \mathrm{H}, \mathrm{br} s, \mathrm{H}-3, \mathrm{~A}+\mathrm{B}), 5.66-5.88(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-4+\mathrm{H}-5, \mathrm{~A}+\mathrm{B}) \mathrm{ppm} . \delta_{\mathrm{C}}(100$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.3\left(\mathrm{CH}_{3}, \mathrm{~A}\right), 14.4\left(\mathrm{CH}_{3}, \mathrm{~B}\right), 42.2(\mathrm{C}-6, \mathrm{~A}), 43.6(\mathrm{C}-6, \mathrm{~B}), 55.9(\mathrm{C}-3, \mathrm{~A}), 56.9(\mathrm{C}-3, \mathrm{~B})$, 61.9 (C-3', $\mathrm{A}+\mathrm{B}), 62.6,62.7,62.8,62.9\left(\mathrm{CH}_{2}, \mathrm{~A}+\mathrm{B}\right), 123.4,124.2,124,6,125.2(\mathrm{C}-4$ or $\mathrm{C}-5, \mathrm{~A}+\mathrm{B})$, 154.9, 155.7, 156.2, 156.3 ( $\mathrm{C}=\mathrm{O}, \mathrm{A}+\mathrm{B}$ ) ppm. HRMS (ESI): calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{NaO}_{5}$ : 281.1108; found: 281.1109. Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{5}$ : C, 51.16; $\mathrm{H}, 6.97$; $\mathrm{N}, 10.85$. Found: $\mathrm{C}, 51.21 ; \mathrm{H}$, 6.91; N, 10.81.

## Synthesis (S)-(1,2,3,6-tetrahydropyridazin-3-yl)methanol (3)

To a stirred solution of cycloadduct $\mathbf{2}(0.119 \mathrm{~g} ; 0.461 \mathrm{mmol})$ in THF ( 2 mL ) was added 2 ml of 2 M NaOH . The mixture was left under reflux for 3 h . After cooling down THF ( 2 ml ) and a suspension of Amberlite $\left(\mathrm{H}^{+}\right)$in water was added to the reaction mixture. The mixture was swirled to promote contact and then immediately filtrated under vacuum. The solvent was evaporated. By adding acetone, water was co-evaporated in the rotary evaporator giving yellow oil ( $0.048 \mathrm{~g} ; 91.0 \%$ ). $[\alpha]_{\mathrm{D}}{ }^{20}-20$ (conc. $0.3 \%$ in EtOH). $\mathrm{v}_{\max }$ (neat): 3422 ( $\mathrm{N}-\mathrm{H}$ ), 1643 ( $\mathrm{C}=\mathrm{C}$ ) $\mathrm{cm}^{-1}$. $\delta_{H}\left(400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 6.01(1 \mathrm{H}, \mathrm{ddd}, J=2.4,5.6,10.4 \mathrm{~Hz} \mathrm{H}-5), 5.79(1 \mathrm{H}, \mathrm{ddd}, \mathrm{J}=2.0,4.4,10.4 \mathrm{~Hz}, \mathrm{H}-$ 4), 3.65-3.57 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3^{\prime}$ ), 3.54-3.48 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3$ ), $3.35(1 \mathrm{H}, \mathrm{ddd}, \mathrm{J}=2.8,5.2,17.6 \mathrm{~Hz} \mathrm{H}-6), 3.24$ ( $1 \mathrm{H}, \mathrm{ddd}, J=2.6,3.2,17.2 \mathrm{~Hz}, \mathrm{H}-6$ ) ppm. $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 127.4$ (C-4 or C-5), 125.0 (C-5 or C4), 62.5 (C-3'), 54.8 (C-3), 44.2 (C-6) ppm. HRMS (ESI): calcd for $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$114.0793; found: $[\mathrm{M}+\mathrm{H}]^{+}$115.0165. Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}$ : C, $52.63 ; \mathrm{H}, 8.77 ; \mathrm{N}, 24.56$. Found: C, 52.67; H, 8.74; N, 24.51.
Synthesis of tert-butyl (S)-3-(hydroxymethyl)-3,6-dihydropyridazine-1(2H)-carboxylate ( N -Boc-1-azafagomine) (4)

To a stirred solution of compound 3 ( $0.044 \mathrm{~g} ; 0.385 \mathrm{mmol}$ ) dissolved in ethanol ( 10 mL ) was added solid $\mathrm{NaOH}(0.015 \mathrm{~g} ; 0.385 \mathrm{mmol})$ and $\mathrm{Boc}_{2} \mathrm{O}$ ( $0.015 \mathrm{~g} ; 0.385 \mathrm{mmol}$ ). The reaction
mixture was left under magnetic stirring for 1 h at room temperature. Then the mixture was filtrated through a pad of celite and washed thoroughly with ethanol. The solution was concentrated to give a yellow solid. The yellow solid was dissolved in acetone, filtrated and the solution evaporated to give a yellow oil ( $0.077 \mathrm{~g} ; 93 \%$ ) $[\alpha]_{D}{ }^{20}+670$ (conc. $0.78 \%$ in DCM). $\mathrm{v}_{\text {max }}$ (neat): $3269(\mathrm{OH}), 1694(\mathrm{CO}) \mathrm{cm}^{-1} . \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 5.96(1 \mathrm{H}, \mathrm{brd}, J=8.8, \mathrm{H}-5), 5.90(1 \mathrm{H}$, bd, $J=8.8, H-4), 4.09(1 \mathrm{H}, \mathrm{bd}, J=10.0 \mathrm{~Hz}, \mathrm{H}-6), 3.96(1 \mathrm{H}, \mathrm{bd}, \mathrm{J}=10.0 \mathrm{~Hz}, \mathrm{H}-6), 3.68-3.56(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ $\left.3, \mathrm{H}-3^{\prime}\right), 1.50\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right) \mathrm{ppm} . \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right), 156.8(\mathrm{C}=\mathrm{O}), 125.1,125.0(\mathrm{C}-4$ and $\mathrm{C}-5)$, 82.7 (C), $61.8\left(\mathrm{C}-3^{\prime}\right), 55.9(\mathrm{C}-3), 27.6\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$ ppm. HRMS (ESI): calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$ 214.1317; found: $[\mathrm{M}-\mathrm{H}]^{-}$213.1591. Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 56.07; H, 8.41; N, 13.08. Found: C, 56.11; H, 8.37; N, 13.05.

Synthesis of tert-butyl (S)-7-oxo-4a,5-dihydro-7H-oxazolo[3,4-b]pyridazine-1(2H)-carboxylate (5)

To a stirred solution of compound $4(0.114 \mathrm{~g} ; 0.532 \mathrm{mmol})$ in dry DCM ( 10 ml ) was added DIPEA (10 equiv.; $0.93 \mathrm{~mL} ; 0.532 \mathrm{mmol}$ ) and triphosgene ( 0.5 equiv.; $0.077 \mathrm{~g} ; 0.532 \mathrm{mmol}$ ). The mixture was left stirring under nitrogen for 5 min at room temperature. The solvent was removed, and the residue was purified by column chromatography (EtOAc : Pet. ether, $1: 1$ ) to give yellow oil ( 0.077 g ; 93\%). [ $\alpha]_{\mathrm{D}}{ }^{20}+30$ (conc. $0.4 \% \mathrm{DCM}$ ). $\mathrm{v}_{\max }$ (neat): 1717 ( $\left.\mathrm{C}=0\right)_{\mathrm{cm}} \mathrm{cm}^{-1} . \delta_{H}$ ( 400 $\mathrm{MHz}, \mathrm{DMSO}) 5.89(1 \mathrm{H}, \mathrm{dm}, J 8.8) 5.86(1 \mathrm{H},(1 \mathrm{H}, \mathrm{ddd}, \mathrm{J}=8.4,4.8,2.0 \mathrm{~Hz}), 4.02(2 \mathrm{H}, \mathrm{dd}, \mathrm{J}=4.8,2.4$ Hz ), $3.71(1 \mathrm{H}, \mathrm{br} \mathrm{s}), 3.55(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=11.2,4.8 \mathrm{~Hz}), 3.46(1 \mathrm{H}, \mathrm{J}=11.2,7.2 \mathrm{~Hz}) \mathrm{ppm} . \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$, DMSO-d ${ }_{6}$ ), 153.6 (CO), $152.6(\mathrm{CO}), 124.5-123.8(\mathrm{C}-\mathrm{H}, \mathrm{Ar}), 83.0,81.6(\mathrm{Cq}), 60.4,59.9\left(\mathrm{CH}_{2} \mathrm{O}\right)$, 56.2, 55.6, $53.0(\mathrm{CH}), 43.8,43.3\left(\mathrm{CH}_{2} \mathrm{~N}\right), 27.9,27.8,27.7,27.4\left(\mathrm{CH}_{3}\right)$ ppm. HRMS (ESI): calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}[\mathrm{M}+\mathrm{H}]^{+} 240.0793$; found: $[\mathrm{M}+\mathrm{H}]^{+}$241.0165. Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}$ : C, 55.00; H , 6.66 ; N, 11.66. Found: C, 55.05; H, 6.63; N, 11.69.

Synthesis of (3S,4R,4aR)-tert-butyl 3,4-dihydroxy-7-oxohexahydro-1H-oxazolo[3,4-

## b]pyridazine-1-carboxylate (6)

Compound 5 ( $0.076 ; 0.316 \mathrm{mmol})$ was dissolved in a mixture of acetone ( 1.2 ml ) and water ( 0.11 ml ). To this solution $N$-methylmorpholine $N$-oxide ( 1.3 equiv., $0.048 \mathrm{~g} ; 0.316 \mathrm{mmol}$ ) followed by a solution of $\mathrm{OsO}_{4}$ in water ( $4 \% ; 34 \mu \mathrm{l} ; 0.0053 \mathrm{mmol}$ ) were added. The reaction mixture was stirred for 1 day at room temperature. The solvents were evaporated until a residue, which was subsequently dissolved in ethyl acetate ( 20 ml ) and washed with water ( 2 x 15 ml ). The organic fractions were combined, dried over $\mathrm{MgSO}_{4}$, filtrated, and the solution concentrated in the rotary evaporator to give an orange oil ( $0.078 \mathrm{~g} ; 90 \%$ ). $v_{\max }$ (neat): $1693(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} . \mathrm{NMR} \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right), 1.5\left(9 \mathrm{H}, \mathrm{s}, 3 \mathrm{CH}_{3}\right), 3.42(1 \mathrm{H}, \mathrm{m}), 3.6(2 \mathrm{H}, \mathrm{s}, \mathrm{OH}), 3.81$
$(2 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}), 3.91(2 \mathrm{H}, \mathrm{q}, J=3.6 \mathrm{~Hz}), 4.34(2 \mathrm{H}, \mathrm{t}, J=2 \mathrm{~Hz}) . \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}\right), 28.07\left(3 \mathrm{CH}_{3}\right)$, $29.6\left(\mathrm{CH}_{2} \mathrm{~N}\right), 36.5(\mathrm{CHN}), 65.1\left(\mathrm{CH}_{2} \mathrm{O}\right), 66.4(\mathrm{CH}-\mathrm{OH}), 68.8(\mathrm{CH}-\mathrm{OH}), 82.6(\mathrm{C}), 155.8(\mathrm{CO}), 162.6$ (CO) ppm. HRMS (EI): m/z calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{6}$ : $274.1165[\mathrm{M}+\mathrm{H}]^{+}$, found: $[\mathrm{M}+\mathrm{H}]^{+} 274.0658$. Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{6}$ : C, 48.17; $\mathrm{H}, 6.56 ; \mathrm{N}, 10.21$. Found: C, 48.21; $\mathrm{H}, 6.59 ; \mathrm{N}, 10.17$.

## Synthesis of (3S,4R,4aR)-3,4-dihydroxytetrahydro-1H-oxazolo[3,4-b]pyridazin-7(2H)-one (7)

To a solution of compound $6(0.099 \mathrm{~g}, 0.361 \mathrm{mmol})$ in acetonitrile ( 3 ml ) was added $\mathrm{H}_{2} \mathrm{O}$ $(11 \mathrm{ml})$ and $\mathrm{Bil}_{3}(0.199 \mathrm{~g}, 0.726 \mathrm{mmol})$. The mixture was refluxed for 16 h (water bath $\left.110{ }^{\circ} \mathrm{C}\right) . \mathrm{A}$ column chromatography (DCM : EtOH, 7:3) was run giving a brown oil ( $0.069 \mathrm{~g} ; 0.396 \mathrm{mmol}, 55$ \%). $v_{\max }$ (neat): $3404(\mathrm{OH}), 1745(\mathrm{CO}) \mathrm{cm}^{-1}$. ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta_{H} 4.57(1 \mathrm{H}, \mathrm{br} \mathrm{s}), 4.44(1 \mathrm{H}$, dd, $\left.J=7.6,8.8 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 4.26\left(1 \mathrm{H}, \mathrm{dd}, J=3.2,8.8 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 3.86(1 \mathrm{H}, \mathrm{ddd}, \mathrm{J}=3.6,7,6,10 \mathrm{~Hz}, \mathrm{H}-5), 3.83-$ $3.81(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.51(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=2.6,9.8 \mathrm{~Hz}, \mathrm{H}-4), 2.97(1 \mathrm{H}, \mathrm{dd}, J=2.8,14.0 \mathrm{~Hz}, \mathrm{H}-6), 2.90$ ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=1.2,13.2 \mathrm{~Hz}, \mathrm{H}-6$ ) ppm. $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 159.2$ (CO), 70.7 (C-4), 67.0 (C-3), $66.5\left(\mathrm{CH}_{2} \mathrm{O}\right)$, $55.0(\mathrm{C}-5)$, $52.5\left(\mathrm{CH}_{2}-6\right) \mathrm{ppm} . \mathrm{HRMS}(\mathrm{ESI}): \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{NaO}_{4}: 197.0538$ [M $+\mathrm{Na}]^{+}$, found: $[\mathrm{M}+\mathrm{Na}]^{+}$197.0533. Anal. Calcd. for $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{4}: \mathrm{C}, 41.37 ; \mathrm{H}, 5.74 ; \mathrm{N}, 16.09$. Found: C, 41.39; H, 5.71; N, 16.11.

## Biological activity

Antibacterial activity of compounds 3-7 and pristine antibiotics (cefotaxime and ceftriaxone) were tested by diffusion method on Staphylococcus aureus and Escherichia coli, as described by Mayrhofer [47]. Pristine antibiotics (cefotaxime and ceftriaxone) were taken in amount $25 \mu \mathrm{~g}$ (indicator disks were purchased by Research-and-Development Center of Pharmacotherapy, 192236 St. Petersburg). The synthesized substances were also taken in the amount equal to $25 \mu \mathrm{~g}$. Escherichia coli was cultivated on Endo's medium and Staphylococcus aureus on Baird-Parker agar. Microbiological tests were performed on Petri dishes. Due to the fact that this method provides only quality data, microdilution method was also performed, as it is written by Jorgensen and Lee [48]. By this method, the MICs of the prepared compounds 37 and usual antibiotics were identified and compared to each other. In order to perform the microdilution method, the stock solutions with different concentrations of the substances were prepared in distilled sterile water and were distributed in 96 multi-well plates. Each well was inoculated with 0.1 ml of microbial suspensions of 0.5 Mc Farland turbidity, prepared from 24 h fresh culture. Sterility control wells (nutrient broth) and microbial growth controls (inoculated nutrient broth) were used. The plates were incubated for 24 h at $37^{\circ} \mathrm{C}$.

## REFERENCES

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Figure S1. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound 1


Figure S2. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound $\mathbf{1}$


Figure S3. dept $135 \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound $\mathbf{1}$


Figure S4. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound 2


Figure S5. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right)$ spectrum of compound 3


Figure S6. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right)$ spectrum of compound 3


Figure S7. dept 135 NMR $\left(D_{2} \mathrm{O}\right)$ spectrum of compound $\mathbf{3}$


Figure S8. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right)$ spectrum of compound 4


Figure S10. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ) spectrum of compound 5


Figure S11. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{DMSO}_{\mathrm{d}} \mathrm{d}_{6}\right)$ spectrum of compound 5


Figure S12. dept 135 NMR $\left(\right.$ DMSO- $\left._{6}\right)$ spectrum of compound 5


Figure S13. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound 6


Figure S14. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound 6


Figure S15. dept $135 \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ spectrum of compound 6


Figure S16. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$ spectrum of compound 7


Figure S17. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$ spectrum of compound 7



Figure S18. dept 135 NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$ spectrum of compound 7


Figure S19. IR (ATR) spectrum of compound 2


Figure S20. IR (ATR) spectrum of compound $\mathbf{3}$


Figure S21. IR (ATR) spectrum of compound 4


Figure S22. IR (ATR) spectrum of compound 5


Figure S23. IR (ATR) spectrum of compound 6


Figure S24. IR (ATR) spectrum of compound 7


Figure S25. Mass spectra of compound $\mathbf{3}$


Figure S26. Mass spectra of compound 4


Figure S27. Mass spectra of compound 6


Figure S28. Mass spectra of compound 7

