



Universidade do Minho Escola de Engenharia



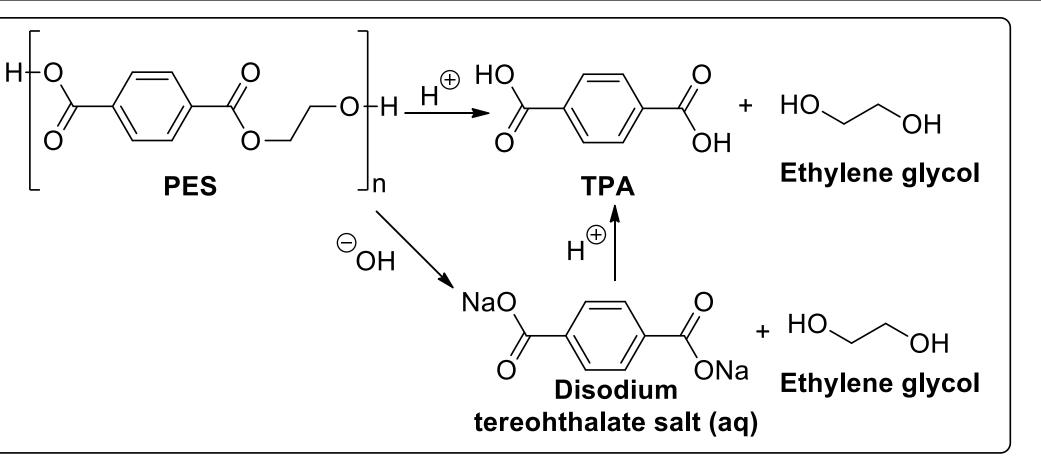
Recovery and recycling of cotton and cotton-containing garments

Bárbara Vieira¹, Marlene Rocha¹, <u>Fábio Pedroso de Lima¹</u>, Rita Gomes-Dias¹, Talita Nicolau¹, Cátia Alves¹, Jorge Padrão¹, Andrea Zille^{1,*}

> ¹Centre for Textile Science and Technology (2C2T), University of Minho, Guimarães, Portugal azille@det.uminho.pt

Textile waste represent an abundant source of synthetic and natural fibres (nearly 92 Mt year-) that must be reused to mitigate the environmental impact of the petrochemical extraction, as well as an intensive use of chemicals and water requirements. In this work, cotton (CO) recovery was tested

Introduction





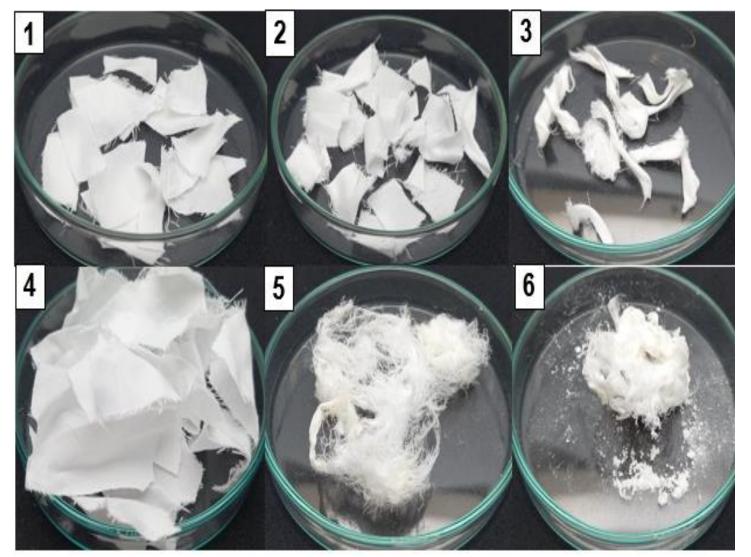
through selective removal of polyester (PES) from the mixture and subsequent terephthalic acid (TPA) isolation (Scheme 1), using post-consumer polycotton textile waste (Figure 1).

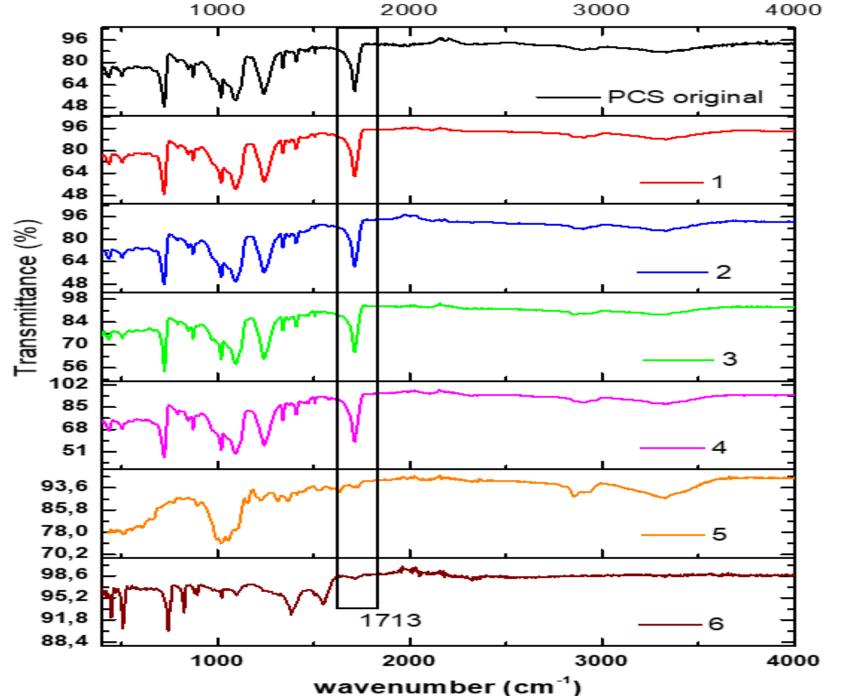
Scheme 1. Schematic representation of the acidic and alkaline hydrolysis of PES and respective products

Figure 1. Pictures of the original materials. From left to right: PCS (75% PES/23% CO/2% EL) and SQUASH POPLIN (84% CO/14% PES/2% EL)

Optimization of PES Hydrolysis

PES removal was optimized using different co-solvents for 24 hours, bath ratio of 1:50 and 50 °C (Figure 2). The best results were obtained when NaOH 2M/DMSO or NaOH 3M/isopropanol. This was confirmed by the disappearance of the C=O stretching signal at 1713 cm⁻¹ characteristic of PES fibre (Figure 2). 2000 3000





PES removal cycles

The best results were obtained using NaOH 2M with DMSO or iso-propanol. Hydrolysis cycles were performed to evaluate the mixtures' capacity for PES removal in successive uses, displaying a maximum yield of 3 cycles, with nearly 100% of CO recovery (Figure 3A and **B**). The removal of PES was followed by FTIR ATR spectroscopy (figures 3C and **D**), targeting the absence of the 1713 cm⁻¹ signal.

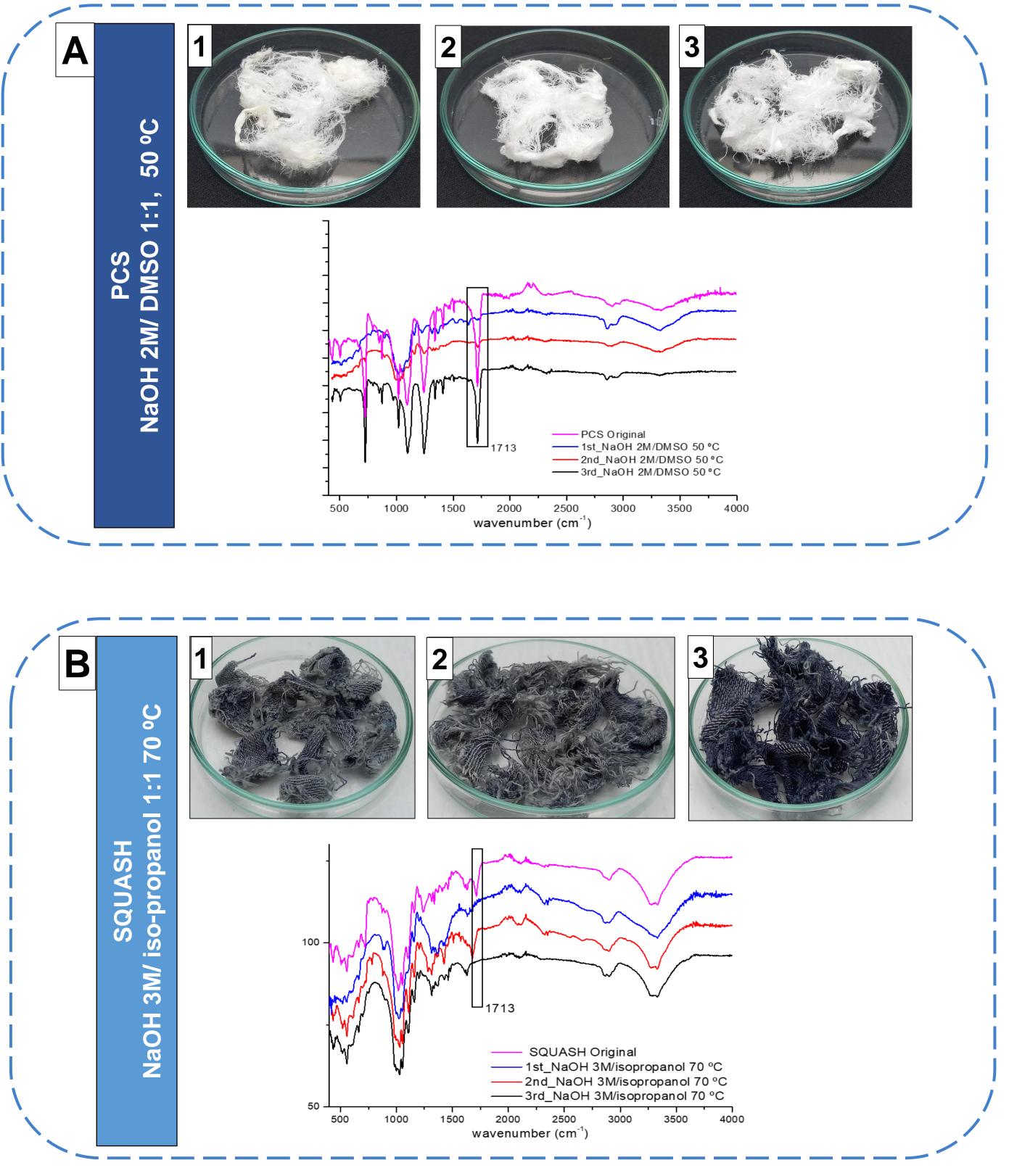


Figure 2. Pictures of a selection of the recovered material after optimization of the reaction conditions, within a fixed time of 24 hours, bath ratio of 1:50 and 50 °C (left), and superimposition of their respective FTIR ATR spectra (1) NaOH 2M; (2) NaOH 2M, DMSO 10% (v/v); (3) NaOH 2M/acetone (1:1); (4) NaOH 2M/HFIP (1:1); (5) NaOH 2M/DMSO (1:1); (6) NaOH 3M/isopropanol (1:1)

TPA recovery

For TPA recovery. the laundering mixture was acidified to pH= 2 and stirred at 50 °C for 2 hours, then cooled to 0 °C and filtered under vacuum. TPA recovery denoted moderate to high yields, with high degree of purity as confirmed by ¹H NMR (**Figure 4**).

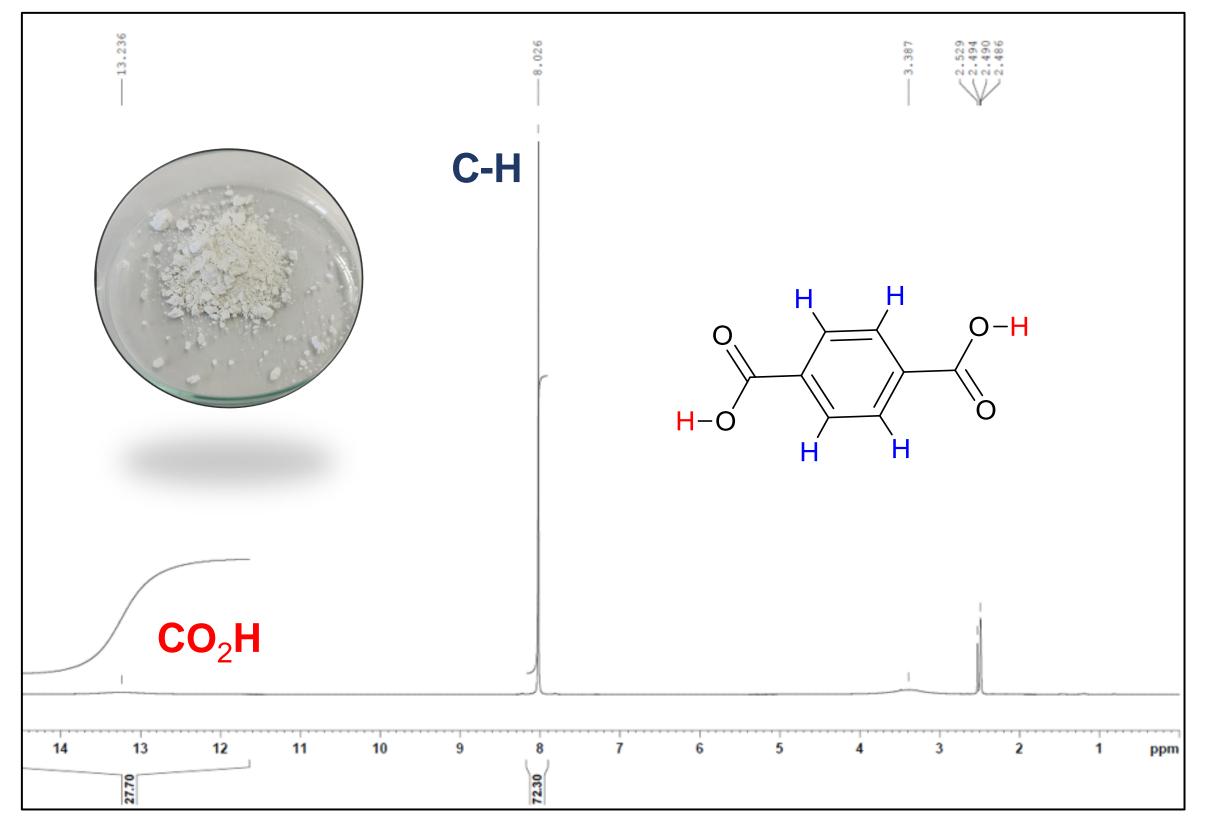


Figure 3. Pictures of a selection of the recovered material after successive hydrolysis cycles, from 1st to 3rd cycle, and respective FTIR ATR spectra; (A) PCS; (B) SQUASH

Figure 4. Example of recovered TPA from the hydrolysis of PET using NaOH 2M/DMSO 1:1, 50 °C, Bath ratio of 1:50, for 24h (top); and its respective ¹H NMR spectrum (400 MHz, DMSO-d₆; bottom)

Aknowledgments

This research was funded by FEDER funds through the Operational Competitiveness Program-COMPETE, under the project POCI-01-0247-FEDER-047124, and by National Funds through Fundação para a Ciência e Tecnologia (FCT), under the project UID/CTM/00264/2020. Talita Nicolau and Cátia Alves acknowledge FCT, MCTES, FSE and UE PhD grants 2022.15386.BD and 2022.10454.BD, respectively.

Conclusions

- ✓ A new and effective method for the recovery of CO and removal of PES was achieved in
 - high yields regarding both cotton recovery and polyester degradation.
- \checkmark The optimized reaction conditions were successfully optimized for usage in tandem hydrolysis cycles up to 3 cycles.
- \checkmark Both the recovered CO and TPA demonstrated good properties that allow a reincorporation into the industrial process.
- \checkmark Further studies are being completed to evaluate the mechanical and physical properties of the recovered CO.









COMPETE

2020



www.2c2t.uminho.pt +351-253 510 289